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Key indicators

Single-crystal X-ray study T = 130 K Mean σ (C–C) = 0.005 Å R factor = 0.045 wR factor = 0.120 Data-to-parameter ratio = 14.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. Trichloro[tris(3-mesitylpyrazol-1-yl)borohydrido]titanium dichloromethane disolvate

In the title complex, $[Ti(C_{36}H_{40}BN_6)Cl_3]\cdot 2CH_2Cl_2$, the Ti atom is coordinated by three N atoms that are constrained to a facial arrangement by the Tp^{Ms} ligand $[Tp^{Ms} = HB(3\text{-mesityl-pyrazolyl})_3]$ and three chloride ligands in a distorted octahedral geometry. The complex molecules are located on crystallographic mirror planes.

Comment

Group 4 metal tris(pyrazolyl)borate complexes of general formula Tp'*M*Cl₃ [Tp' = generic tris(pyrazolyl)borate ligand; M = Ti, Zr, or Hf] can be activated by methylalumoxane (MAO) to generate active catalysts for ethylene polymerization and ethylene/hexene copolymerization (Murtuza *et al.*, 2002; Michiue & Jordan, 2003; Michiue & Jordan, 2004; Gil *et al.* 2004). The activity of Tp'*M*Cl₃/MAO catalysts is strongly influenced by the steric properties of the Tp' ligands, and catalysts that contain the sterically bulky Tp^{Ms} and Tp^{Ms*} ligands [Tp^{Ms} = HB(3-mesitylpyrazolyl)₃; Tp^{Ms*} = HB(3-mesitylpyrazolyl)₂(5-mesityl-pyrazolyl)] exhibit very high activity.



As part of an ongoing investigation of the properties of these catalysts (Lee & Jordan, 2005), we have structurally characterized the title complex, Tp^{Ms}TiCl₃, (I). The molecular structure of (I) is shown in Fig. 1, and selected bond distances and angles are presented in Table 1. The Ti atom is coordinated by three N atoms that are constrained to a facial arrangement by the Tp^{Ms} ligand, and by three chloride ligands in a distorted octahedral geometry. The complex is located on a mirror plane. The core structure of (I) is typical for tris-(pyrazolyl)borate titanium complexes. The N-Ti-N angles formed by the tridentate Tp^{Ms} ligand are acute [range $83.25 (8) - 83.40 (12)^{\circ}$, the Cl-Ti-Cl angles are correspondingly larger [range 94.78 (5)-95.22 (3)°], and the Ti-Cl distances are normal [range 2.2417 (12)-2.2635 (9) Å]. Very similar values for these parameters were observed for Tp^{Ms*}TiCl₃ [(II); Michiue & Jordan, 2003], Tp^{*}TiCl₃ [(III); $Tp^* = HB(3,5-Me_2-pyrazolyl)_3$ (Antiñolo *et al.*, 1999), $Tp^{Np}TiCl_3$ [(IV); $Tp^{Np} = HB(3\text{-neopentyl-pyrazolyl})_3$] (Gil &

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metal-organic papers



View of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms have been omitted. Symmetry-equivalent atoms are generated by the (010) mirror. The suffix A corresponds to the symmetry position (x, -y, z). The solvent molecules have been omitted.

Casagrande, 2004), and Tp^{Menth*}TiCl₃ [(V); Tp^{Menth*} = hydrotris[4,5,6,7-tetrahydro-4-methyl-7-(1-methylethyl)-2*H*indazolato-N2]borate] (LeCloux *et al.*, 1994). The Ti-N distances in (I) [range 2.206 (2)–2.253 (3) Å] are similar to those in the isomeric complex (II) [range 2.172 (3)– 2.239 (3) Å], which also contains bulky mesityl groups, but slightly longer than those in (III) [range 2.164 (2)– 2.177 (2) Å], (IV) [range 2.162 (8)–2.167 (6) Å], and (V) [range 2.131 (7)–2.186 (6) Å], which contain sterically smaller Tp' ligands. In (I), the three mesityl rings lie roughly perpendicular to the pyrazolyl rings to which they are attached and together form a deep pocket that shields the three chloride ligands.

Experimental

The synthesis of (I) was carried out as described previously (Murtuza *et al.*, 2002). Crystals of (I) were grown from a saturated CH_2Cl_2 solution at 273 K.

Z = 2

 $D_{\rm r} = 1.377 \ {\rm Mg \ m^{-3}}$

 $0.45 \times 0.15 \times 0.15 \ \mathrm{mm}$

Mo $K\alpha$ radiation

 $\mu = 0.67 \text{ mm}^{-1}$

T = 130 (2) K

Rod. red

Crystal data

| $[Ti(C_{36}H_{40}BN_6)Cl_3] \cdot 2CH_2Cl_2$ |
|--|
| $M_r = 891.65$ |
| Monoclinic, $P2_1/m$ |
| a = 7.964 (3) Å |
| b = 22.940(7) Å |
| c = 11.844 (4) Å |
| $\beta = 96.484 \ (5)^{\circ}$ |
| $V = 2149.9 (12) \text{ Å}^3$ |
| |

Data collection

Bruker SMART APEX CCD diffractometer ω scans Absorption correction: multi-scan (*SADABS*; Bruker, 2002) $T_{min} = 0.670, T_{max} = 0.90$ 16493 measured reflections 3894 independent reflections 3026 reflections with $I > 2\sigma(I)$ $R_{int} = 0.047$ $\theta_{max} = 25.0^{\circ}$

Refinement

| Refinement on F^2 | $w = 1/[\sigma^2(F_o^2) + (0.0635P)^2]$ |
|---------------------------------|--|
| $R[F^2 > 2\sigma(F^2)] = 0.045$ | + 0.8057P] |
| $wR(F^2) = 0.120$ | where $P = (F_0^2 + 2F_c^2)/3$ |
| S = 1.05 | $(\Delta/\sigma)_{\rm max} = 0.002$ |
| 3894 reflections | $\Delta \rho_{\rm max} = 0.92 \ {\rm e} \ {\rm \AA}^{-3}$ |
| 264 parameters | $\Delta \rho_{\rm min} = -0.64 \text{ e } \text{\AA}^{-3}$ |
| H-atom parameters constrained | |

Table 1

Selected geometric parameters (Å, °).

| Ti1-N1 | 2.253 (3) | Ti1-Cl1 | 2.2635 (9) |
|-------------------------|-------------|--------------------------|-------------|
| Ti1-N3 | 2.206 (2) | Ti1-Cl2 | 2.2417 (12) |
| B1-N2 | 1.539 (6) | B1-N4 | 1.541 (4) |
| N2-N1-Ti1 | 118.3 (2) | Cl2-Ti1-N1 | 172.53 (9) |
| N4-N3-Ti1 | 119.33 (17) | N3 ⁱ -Ti1-Cl1 | 171.27 (6) |
| N3-Ti1-N1 | 83.25 (8) | N3 ⁱ -Ti1-Cl2 | 91.19 (6) |
| N3 ⁱ -Ti1-N3 | 83.40 (12) | Cl1-Ti1-Cl1 ⁱ | 94.78 (5) |
| N3-Ti1-Cl1 | 90.53 (6) | Cl2-Ti1-Cl1 | 95.22 (3) |
| N1-Ti1-Cl1 | 89.83 (6) | N2 - B1 - N4 | 108.1 (2) |
| N3-Ti1-Cl2 | 91.19 (6) | N4 ⁱ -B1-N4 | 108.7 (3) |
| | | | |

Symmetry code: (i) $x, -y + \frac{3}{2}, z$.

Atom H22 attached to atom B1 was refined freely [B-H = 1.13 (4) Å]. All other H atoms were positioned geometrically (C-H = 0.95 and 0.98 Å) and refined as riding, with $U_{iso}(H) = 1.2$ or 1.5 times U_{eq} (parent atom).

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT-Plus* (Bruker, 2002); data reduction: *SADABS* (Bruker, 2002); program(s) used to solve structure: *SHELXTL* (Bruker, 2002); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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